



Elektronegativitätswerte

${}^1\text{H}$ 2,1							${}^2\text{He}$ –
${}^3\text{Li}$ 1,0	${}^4\text{Be}$ 1,5	${}^5\text{B}$ 2,0	${}^6\text{C}$ 2,5	${}^7\text{N}$ 3,0	${}^8\text{O}$ 3,5	${}^9\text{F}$ 4,0	${}^{10}\text{Ne}$ –
${}^{11}\text{Na}$ 0,9	${}^{12}\text{Mg}$ 1,2	${}^{13}\text{Al}$ 1,5	${}^{14}\text{Si}$ 1,8	${}^{15}\text{P}$ 2,1	${}^{16}\text{S}$ 2,5	${}^{17}\text{Cl}$ 3,0	${}^{18}\text{Ar}$ –
${}^{19}\text{K}$ 0,8	${}^{20}\text{Ca}$ 1,0	${}^{31}\text{Ga}$ 1,6	${}^{32}\text{Ge}$ 1,8	${}^{33}\text{As}$ 2,0	${}^{34}\text{Se}$ 2,4	${}^{35}\text{Br}$ 2,8	${}^{36}\text{Kr}$ –
${}^{37}\text{Rb}$ 0,8	${}^{38}\text{Sr}$ 1,0	${}^{49}\text{In}$ 1,7	${}^{50}\text{Sn}$ 1,8	${}^{51}\text{Sb}$ 1,9	${}^{52}\text{Te}$ 2,1	${}^{53}\text{I}$ 2,5	${}^{54}\text{Xe}$ –
${}^{55}\text{Cs}$ 0,7	${}^{56}\text{Ba}$ 0,9	${}^{81}\text{Tl}$ 1,8	${}^{82}\text{Pb}$ 1,9	${}^{83}\text{Bi}$ 1,9	${}^{84}\text{Po}$ 2,0	${}^{85}\text{At}$ 2,2	${}^{86}\text{Rn}$ –
${}^{87}\text{Fr}$ 0,7	${}^{88}\text{Ra}$ 0,9						



Metall



Halbmetall, Nichtmetall oder Edelgas

Redoxreihe der Metalle

Li Cs K Ca Na Y Mg Al Mn Zn Cr Fe Cd Co Ni Sn Pb H_2 Cu Ag Hg Pt Au